

Properties of Atoms, Elements, and Compounds	Unit	Cl	CHECKPOINT		
Properties of Atoms, Elements, and Compounds	Offic	1	2	3	
PS.3 Matter and energy. The student knows that matter is composed of atoms and can differentiate elements and compounds.					
The student will explain what it means to say that God created the world and all matter out of nothing at a certain point in time; how it manifests His wisdom, glory, and purpose; and how He holds everything in existence according to His plan. *					

Catho	olic Identity: Integration of Our Faith
MS.1A	display a deep sense of wonder and delight about the natural universe *
MS.1B	describe the unity of faith and reason *
MS.1C	describe relationships, elements, underlying order, harmony, and meaning *
MS.1D	share concern and care for the environment as part of God's creation *

Logran	ing Propose Standards (Tools to Know)	Unit	CHECKPOINT				
Learn	ing Process Standards (Tools to Know)	Onit	1	2	3		
MS.2A MS.2B	plan and conduct investigations collect information using appropriate scientific tools						

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Conte	ent	Unit	1	2	3		
Structu	ire of Atoms						
PS.3A	describe the structure of atoms, including the masses, electrical charges, and locations, of protons and neutrons in the nucleus and electrons in the electron cloud						
PS.3A.1	identify that protons determine an element's identity and valence electrons determine its chemical properties, including reactivity						
Period	ic Table						
PS.3B	interpret the arrangement of the Periodic Table, including groups and periods, to explain how properties are used to classify elements						
PS.3B.1	know that an element is a pure substance represented by a chemical symbol and that a compound is a pure substance represented by a chemical formula						

Loarr	Process Standards (Ways to Show)	Unit	Cŀ	CHECKPOINT		
Lean	ling Frocess standards (ways to snow)	Onit	1	2	3	
MS.2C	record and organize data and observations					
MS.2D	communicate observations about investigations					

develop models to describe the atomic composition of simple molecules and extended

PS.3C

structures.



Chami	cal Formulas, Equations, and Reactions	Unit	CHECKPOINT				
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PS.3	Matter and energy. The student knows that matter has chemical and physical properties.						

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Conte	ent	Unit	1	2	3
Chemic	cal Formulas				
PS.3D	recognize that chemical formulas are used to identify substances and determine the number of atoms of each element in chemical formulas containing subscripts				
Chemic	cal Reactions				
PS.3E	investigate how evidence of chemical reactions indicates that new substances with different properties are formed and how that relates to the law of conservation of mass				
PS.3E.1	distinguish between physical and chemical changes in matter				
PS.3E.2	identify the formation of a new substance by using the evidence of a possible chemical change such as production of a gas, change in temperature, production of a precipitate, or color change				
PS.3E.3	develop and use a model to describe how the total number of atoms does not change in a chemical reaction and thus mass is conserved				
Chanas	a in Dautiala Matian				
Change	e in Particle Motion				

Chang	e in Particle Motion	
PS.3F	develop a model that predicts and describes changes in particle motion, temperature, and state of a pure substance when thermal energy is added or removed	

Learning Process Standards (Ways to Show)	Unit	CHECKPOINT			
Lean	ling Frocess standards (ways to snow)	Onit	1	2	3
MS.2C	record and organize data and observations				
MS.2D	communicate observations about investigations				



Force, Motion, and Energy	Unit	CHECKPOINT				
Force, Motion, and Energy	Oilit	1	2	3		
PS.4 Force, motion, and energy. The student knows that there is a relationship between force, motion, and energy.						

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MS.2A	plan and conduct investigations						
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Motion					
PS.4A	investigate and describe applications of Newton's three laws of motion				
PS.4A.1	compare and contrast potential and kinetic energy				
Force					
PS.4B	plan an investigation to provide evidence that the change in an object's motion depends on the sum of the forces on the object and the mass of the object				
PS.4B.1	demonstrate and calculate how unbalanced forces change the speed or direction of an object's motion				
PS.4B.2	differentiate between speed, velocity, and acceleration				
PS.4B.3	calculate average speed using distance and time measurements				
PS.4B.4	measure and graph changes in motion				
PS.4B.5	investigate how inclined planes can be used to change the amount of force to move an object				
Newto	n's Third Law				
PS.4C	apply Newton's Third Law to design a solution to a problem involving the motion of two colliding objects				
PS.4C.1	identify and describe the changes in position, direction, and speed of an object when acted upon by unbalanced forces				
Electric	and Magnetic Forces				
PS.4D	ask questions about data to determine the factors that affect the strength of electric and magnetic forces				
PS.4D.1	conduct an investigation and evaluate the experimental design to provide evidence that fields exist between objects exerting forces on each other even though the objects are not in contact				

Loove	sing Propose Standards (Wayerto Show)	Unit	CHECKPOINT				
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l avv of	Concernation of Fueron.	Unit	CHECK	POINT
	Conservation of Energy ce, motion, and energy. The student knows that the Law of Conservation of Energy states	Offic	1 2	3
tha	t energy can neither be created nor destroyed, it just changes form.			
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			CHECK	POINT
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Conte	ent	Unit	CHECK	
			1 2	3
	al and Kinetic Energy			
PS.4E	develop a model to describe that when the arrangement of objects interacting at a distance changes, different amounts of potential energy are stored in the system			
PS.4E.1	construct and interpret graphical displays of data to describe the relationships of kinetic energy to the mass of an object and to the speed of an object			
Transfe	r of Energy			
PS.4F	apply scientific principles to design, construct, and test a device that either minimizes or maximizes thermal energy transfer			
PS.4F.1	investigate methods of thermal energy transfer, including conduction, convection, and radiation			
PS.4F.2	verify through investigations that thermal energy moves in a predictable pattern from warmer to cooler until all the substances attain the same temperature such as an ice cube melting			
PS.4F.3	demonstrate energy transformations such as energy in a flashlight battery changes from chemical energy to electrical energy to light energy			

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construct, use, and present arguments to support the claim that when the kinetic energy of

an object changes, energy is transferred to or from the object

Changes in Kinetic Energy



	Unit	CHECKPOINT				
Waves and their Application	Offic	1	2	3		
PS.4 Force, motion, and energy. The student knows the characteristics and behavior of waves.						

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Come		Unit	1	2	3	
Waves						
PS.4H	use mathematical representations to describe a simple model for waves that includes how the amplitude of a wave is related to the energy in a wave					
PS.4H.1	develop and use a model to describe that waves are reflected, absorbed, or transmitted through various materials					

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